

IMIDAZOLE CARBOXYLATES BY A CLAISEN-TYPE REARRANGEMENT OF AMIDOXIME-PROPIOLATE ADDUCTS¹

Ned D. Heindel

Department of Chemistry, Lehigh University, Bethlehem, Pa., 18015

Maria C. Chun

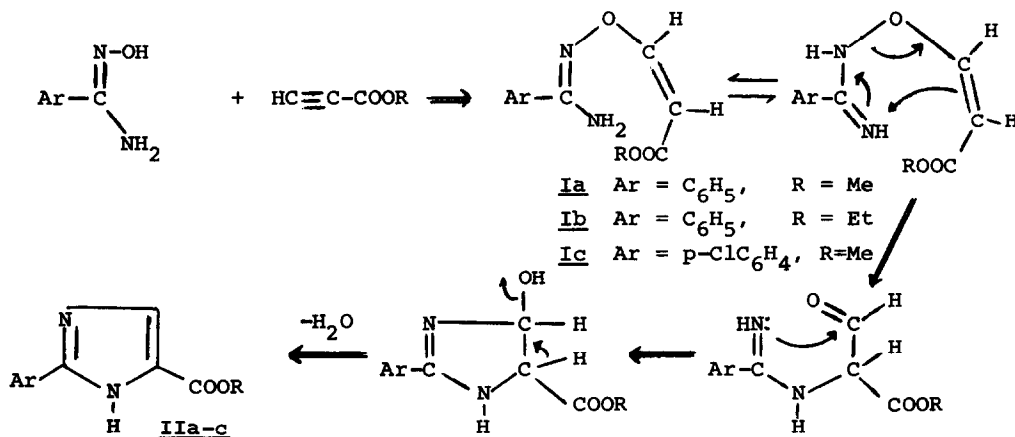
Department of Chemistry, Cedar Crest College, Allentown, Pa., 18104

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Recent publications on Claisen-type rearrangements in nitrogen heterocyclics^{2,3,4} prompt us to report our discovery of a new imidazole carboxylate synthesis which apparently involves the intermediacy of a similar 1,3 sigmatropic shift. Several aromatic amidoximes have been condensed at the alkyne linkage of propiolate esters in 50 to 60% yields by refluxing the components in MeOH for 3 hrs. Since acylation and alkylation studies have shown that the oxygen function of amidoximes is generally the most nucleophilic locus,⁵ and since O-methyl amidoximes did not react with the acetylene ester, we concluded that the adducts represented a transoid addition of OH to C≡C (vicinal coupling $J = 6$ Hz in I a, b, c). Ketoxime addition to dimethyl acetylenedicarboxylate is apparently non-stereospecific.⁶

The adducts — Ia (mp 121 — 123°), Ib (mp 78 — 79°), and Ic (mp 179 — 180°)⁷ — were pyrolyzed by refluxing in phenyl ether for 30 minutes, and the products precipitated by addition of cold hexane. Recrystallization from MeOH yielded 72% IIa (mp 219 — 221°), 70% IIb (mp 188 — 189°)⁸, and 61% IIc (mp 244 — 246°). Elemental and mass spectral analysis confirmed that the products corresponded to the loss of a molecule of water from the starting adducts. Saponification of IIa produced 2-phenyl-4-imidazolecarboxylic acid which was identical with an authentic sample prepared by Fargher and Pyman's method.⁹

An attractive mechanism for this transformation involves a pyrolytic "no mechanism"



rearrangement of the tautomer of the amidoxime adduct through the intermediate amidine, which eluded isolation under the reaction conditions, to the final heterocyclic. Although Sheradsky has proposed a similar rearrangement to account for the formation of pyrroles from ketoxime-dimethyl acetylenedicarboxylate adducts⁶, our example not only represents a unique case of a Claisen-type shift with three hetero atoms in the rearranging cycle but also provides a far more direct synthesis of such imidazole carboxylates than that formerly available.⁹

REFERENCES

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7. Satisfactory analytical data (combustion, NMR, IR,) were obtained for all compounds.
8. Lit. mp for ethyl 2-phenyl-4-imidazolecarboxylate, 187 — 188°, see ref. 9.
9. R. Fargher and L. Pyman, *J. Chem. Soc.*, 217 (1919).